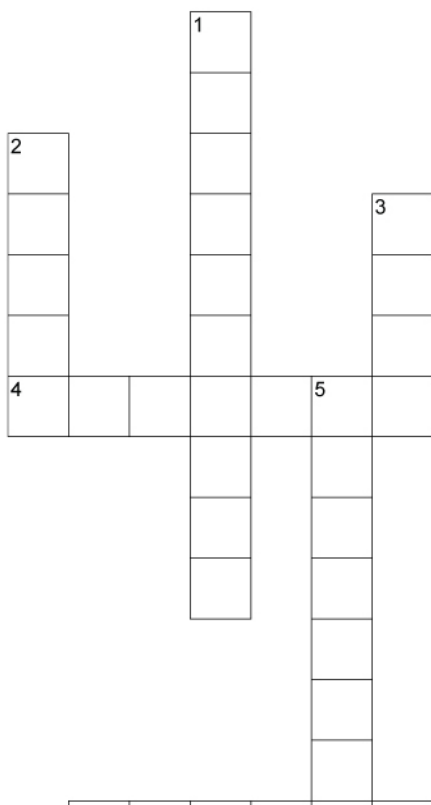
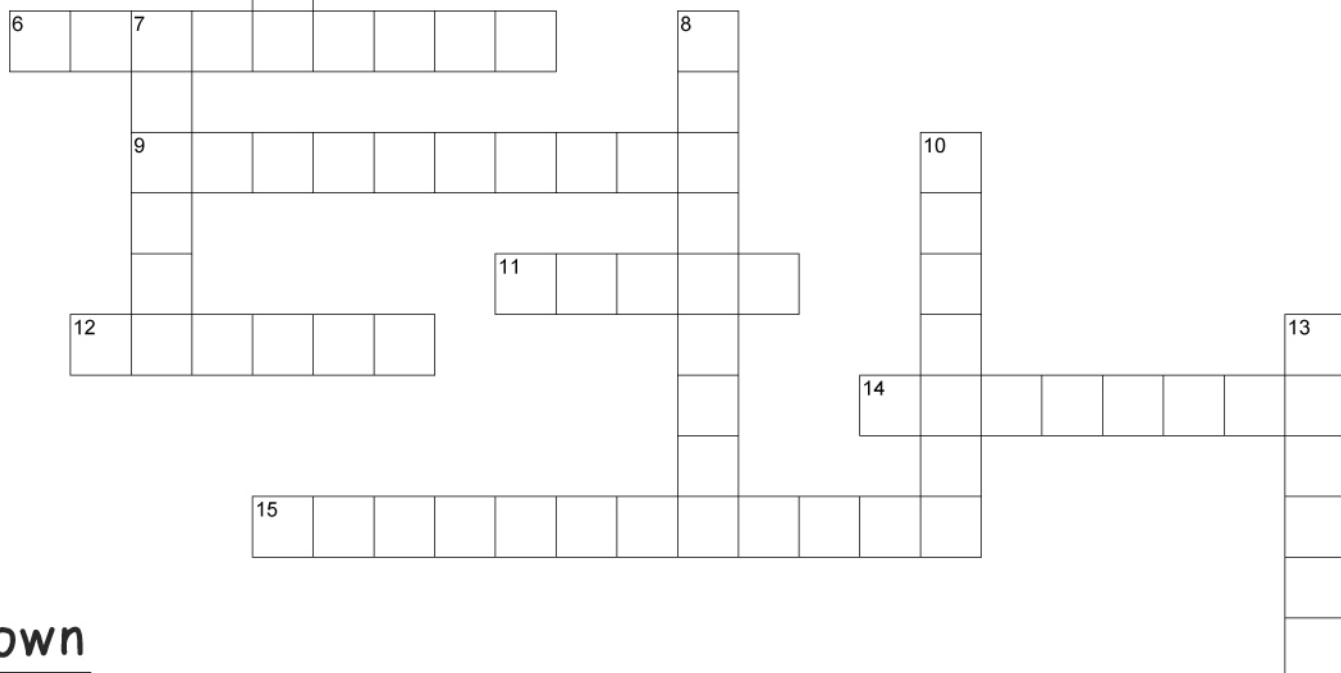


THE PERIODIC TABLE of Elements



Across

4. An element's atomic number is the number of _____ each atom has
6. Giant covalent lattice
9. Elements are classified as metals, non metals, and _____ according to their properties
11. The _____ gases are considered inactive
12. The vertical columns are called _____ or families
14. Metals tend to form ions with a _____ charge
15. Name for the group I metals

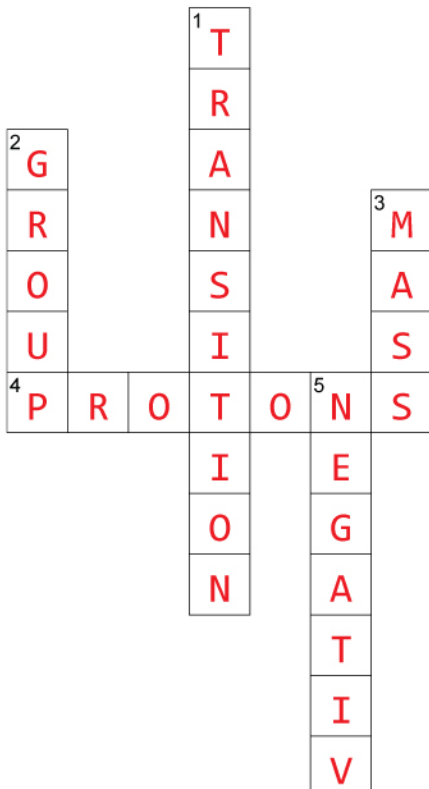


Down

1. The _____ metals are the elements in the middle of the periodic table
2. A column of elements on the periodic table
3. The _____ number is the sum of the protons and neutrons
5. Nonmetals tend to form ions with a _____ charge
7. The atomic _____ is used to arrange the elements
8. _____ metals do not dissolve
10. The horizontal rows are termed _____
13. Most elements are _____

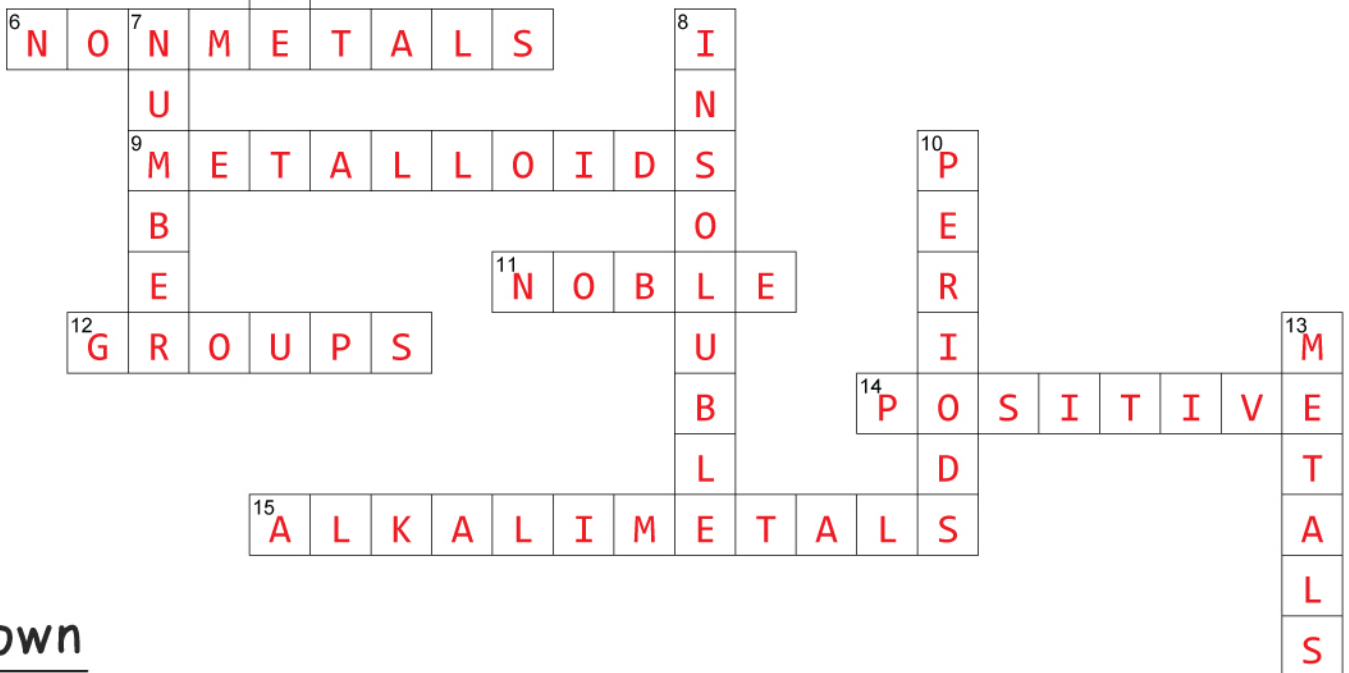
THE PERIODIC TABLE of Elements

(Solution)



Across

- An element's atomic number is the number of _____ each atom has
- Giant covalent lattice
- Elements are classified as metals, non metals, and _____ according to their properties
- The _____ gases are considered inactive
- The vertical columns are called _____ or families
- Metals tend to form ions with a _____ charge
- Name for the group I metals



Down

- The _____ metals are the elements in the middle of the periodic table
- A column of elements on the periodic table
- The _____ number is the sum of the protons and neutrons
- Nonmetals tend to form ions with a _____ charge
- The atomic _____ is used to arrange the elements
- _____ metals do not dissolve
- The horizontal rows are termed _____
- Most elements are _____